A ¹³C NMR CHEMICAL SHIFT STUDY OF trans-FUSED BICYCLIC HEXOPYRANOSIDE DERIVATIVES (Miss) Eileen Conway and R.D. Guthrie

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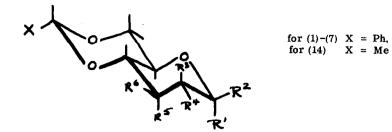
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Natural abundance ¹³C nmr spectroscopy offers a new tool for the study of carbohydrates. Several papers² on free aldoses and their glycosides have shown that structural correlations can be made; only one paper ^{2a} briefly mentioned other derivatives.

As part of a general ¹³C nmr study of monosaccharides and their derivatives we have investigated the spectra of the conformationally more stable, trans-fused methyl 4,6-Q-benzylidene-Daldohexopyranosides and report in this Letter on diols, am ino-alcohols and mono- and di-deoxy derivatives (see Figure). The pyranose ring in these systems has been demonstrated by pmr^3 , ir⁴ and other analytical methods⁵ to be in the Cl(D) conformation, as shown. These compounds therefore enable an investigation of the variations of chemical shift with stereochemical change to be made with confidence.

Noise-decoupled spectra were recorded for thirteen benzylidene compounds and one ethylidene derivative (14). The assignment of δ values (see Table) is based on application of chemical shift theory^{2, 6-8}, and spectral comparison of structurally related compounds. Single-frequency off-resonance spectra of compounds (1), (6) and (12) were used to differentiate between carbons of close chemical shifts. The essentially constant shifts of the aromatic carbons of compounds (1) - (13) were 136.8 (substituted carbon), 128.7 (para), 127.8 and 125.8 ppm.



| Compound (1) | $\frac{\text{Hexose}}{\text{Configuration}} \alpha - \underline{\text{allo}}$ | $\frac{\mathbf{R}^{1}}{\mathbf{OMe}}$ | $\frac{R^2}{H}$ | $\frac{R^3}{H}$ | <u>R</u> ⁴ OH | <u>R</u> OH | R ⁶ H |
|-----------------|---|---------------------------------------|-----------------|-----------------|------------------|----------------|---------------------|
| (2) | a -altro | ОМе | н | он | H | OH | н |
| (3) | β-altro | н | OMe | он | н | ЮК | ÷٢ |
| (4) | a - gluco | ОМе | н | н | ОН | н | он |
| (5) | β - gluco | н | ОМе | н | OH | н | OH |
| (6) | a - manno | ОМе | н | он | н | н | OH |
| (7) | 2, 3-dideoxy-α- erythro | ОМе | н | н | н | н | н |
| (8) | 3-deoxy-α- <u>arabin</u> o | OMe | н | OH | н | н | н |
| (9) | 3-deoxy- <i>a</i> - <u>ribo</u> | ОМе | н | н | ОН | н | н |
| (10) | a-gluco | OMe | н | н | он | н | NH2 |
| (11) | altro altro | OMe | Н | он | Н | NH2 | н |
| (12) | Q-allo | ОМе | H | H | OH | NH2 | н |
| (13) | α-manno | OMe | н | он | н | н | NH2 |
| (14) | β-gluco | н | ОМе | н | ОН | н | OH |

Figure

| ¹³ C Nmr Shifts ^a of Compounds (1) - (14) | | | | | | | | | | | | |
|---|--------------------|------|------|------|------|------|--------------------|------|--|--|--|--|
| Compound | C-1 | C-2 | C-3 | C-4 | C-5 | C-6 | C-7 | ОМе | | | | |
| 1 | 100.2 | 67.9 | 68.8 | 78,1 | 56.9 | 68.8 | 101.5 | 55.7 | | | | |
| 2 | 101.6 | 69.6 | 68.8 | 76.0 | 57.8 | 68.8 | 101.8 | 55.0 | | | | |
| 3 | 99.4 | 70.8 | 68.6 | 76.5 | 62.8 | 68.6 | 101.8 | 56.4 | | | | |
| 4 | 99.9 | 72.4 | 70.5 | 80.8 | 62.0 | 68.5 | 101.5 | 54.9 | | | | |
| 5 | 104.2 | 74.2 | 72.9 | 80.3 | 65.9 | 68.3 | 101.5 | 56.8 | | | | |
| 6 | 101.7 | 70.6 | 68.0 | 78.5 | 62.9 | 68.4 | 101.7 | 54.4 | | | | |
| 7 | 98.1 | 29.0 | 23.5 | 77.9 | 64.5 | 69.1 | 101.6 | 54.2 | | | | |
| 8 | 100.5 | 67.4 | 31.4 | 73.5 | 64.6 | 68.9 | 101.8 | 54.3 | | | | |
| 9 | 98.9 | 67.0 | 32.6 | 76.1 | 63.4 | 68.9 | 101.3 | 54.6 | | | | |
| 10 | 99.4 | 72.1 | 52.4 | 81.0 | 62.2 | 68.7 | 101.5 | 55.0 | | | | |
| 11 | 101.6 ^b | 69.4 | 52.0 | 75.9 | 57.6 | 68.8 | 101.8 ^b | 54.8 | | | | |
| 12 | 100.8 | 67.5 | 52.1 | 78.3 | 56.7 | 68.9 | 101.3 | 55.6 | | | | |
| 13 | 102.0 | 70.0 | 50.3 | 79.4 | 63.4 | 68.5 | 101.6 | 54.5 | | | | |
| 14 [°] | 104.1 | 74.2 | 72.9 | 79.8 | 65.9 | 67.7 | 99.4 | 56.7 | | | | |

TABLE

^aSpectra of 4:1 deuteriochloroform-methanol solutions determined on a Fourier transform spectrometer at 15.08 MHz; chemical shifts in parts per million downfield from TMS; $\delta^{\text{TMS}} = \delta^{\text{CDCl}_3} + 76.9 \text{ ppm}$. ^b Values may be interchanged.

 $^{\rm c}$ $_{\wedge}({\rm C-7~Me}) = 19.6 \text{ ppm}$

Acetalation of the C-4 and C-6 oxygens of the aldohexopyranosides leads to expected deshielding of C-4 and C-6 compared with the unsubstituted glycosides². Being unaffected by stereochemical changes, C-6 and C-7 varied little in chemical shift through the whole series. The assignment of C-6 was confirmed from the spectrum of the 6, 6'- \underline{d}_{g} analogue of compound (4); this spectrum was identical to that of (4) except that the signal at 68.5 ppm was missing.

The C-5 signal of all fourteen compounds was dramatically upfield of its position in unsubstituted glycosides². While greater in magnitude in the present series this shift is analogous to the observed effect of ring oxygens on C-.2 and C-4 of aldohexopyranosides^{2b} and on C-5 of 1, 3-dioxanes⁹. The greater effect in (1)-(14) may be due to their rigid stereochemistry, relative to the analogies quoted. Substituent parameters of the acetals are nearly identical with those of the methyl aldohexopyranosides, as indicated by the effects on C-1 and C-2.

REFERENCEŞ

- Part (X) in a series on ¹³C nmr analysis by G.L. et al. [Part (IX) G. Lukacs, S.D. Gero, D.A. van Dorp, E.W. Hagaman and E. Wenkert, <u>J.Amer.Chem.Soc</u>., in the press].
- (a) L.D. Hall and L.F. Johnson, Chem. Comm., 1969, 509; (b) A.S. Perlin, et al, <u>Tetrahedron Letters</u>, 1969, 2921; Canad. J.Chem., 1970, 48, 2596, 2639; Carbohydrate Res., 1970, 15, 403; (c) D.E. Dorman and J.D. Roberts, J.Amer.Chem.Soc., 1970, 92, 1355; (d) W. Voelter, E. Breitmaier, G. Jung, et al, Angew Chem.Int.Ed., 1971, 10, 935; ⁵Chimica, 1971, 25, 168, ³362; 1972, 26, ⁴136, Chem.Ber., 1971, 104, 1147; (e) D.Doddrell and A. Allerhand, J.Amer.Chem.Soc., 1971, 93, 2777.
- 3 R.D. Guthrie, et al. J. Chem. Soc., 1961, 4166, <u>Tetrahedron</u>, 1968, 24, 4517, <u>Carbohydrate Res.</u>, 1969, 9, 99; B. Coxon, <u>Tetrahedron</u>, 1965, 21, 3481.
- 4 H. Spedding, Adv. Carbohydrate Chem., 1964, 19, 23.
- 5 R.D. Guthrie, et al, <u>J. Chem. Soc. (C)</u>, 1967, 1194, <u>Tetrahedron</u>, 1970, <u>26</u>, 3653, C <u>Carbohydrate Res.</u>, 1969, <u>11</u>, 53, 565; 1970, <u>13</u>, 199; 1971, <u>18</u>, 373.
- 6 G.C. Levy and G.L. Nelson, "Carbon-13 Nuclear Magnetic Resonance for Organic Chemists", Wiley-Interscience, 1972.
- 7 D.K. Dalling and D.M. Grant, J.Amer. Chem. Soc., 1967, 89, 6612.
- 8 D.W. Cochran, Ph.D. Dissertation, Indiana University, 1971.
- 9 A.J. Jones, E.L. Eliel, D.M.Grant, M.C. Knoeber, and W.F. Bailey, <u>J.Amer.Chem.Soc.</u>, 1971, 93, 4772.